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## CS/B.PHARM (NEW)/SEM-3/PT-304/2010-11

# 2010-11 PHARMACEUTICAL CHEMISTRY (ORGANIC)

Time Allotted : 3 Hours

Full Marks : 70

The figures in the margin indicate full marks. Candidates are required to give their answers in their own words as far as practicable.

## **GROUP – A**

## (Multiple Choice Type Questions)

1. Choose the correct alternatives for any *ten* of the following :

 $10 \times 1 = 10$ 

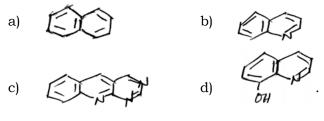
i) The linkage in cellulose is

- a)  $\beta$ -1, 4-linkage b)  $\alpha$ -1, 4-linkage
- c)  $\beta$ -1, 6-linkage d)  $\alpha$ -1, 6-linkage.
- ii) Arrange according to their preference for naming the following heteroatoms :
  - a) O>S>Se>N>P>Si>B b) O>Se>S>P>N>Si>B
  - c) N>P>Si>S>Se>B>O d) O>S>Se>P>N>Si>B.

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- Reaction by which shortening of carbon chain is carried out is known as
  - a) Killiani-Fischer synthesis
  - b) Ruff degradation
  - c) Fischer-Indole synthesis
  - d) Sowden-Fischer synthesis.
- iv) Which of the following is not a heterocyclic compound ?



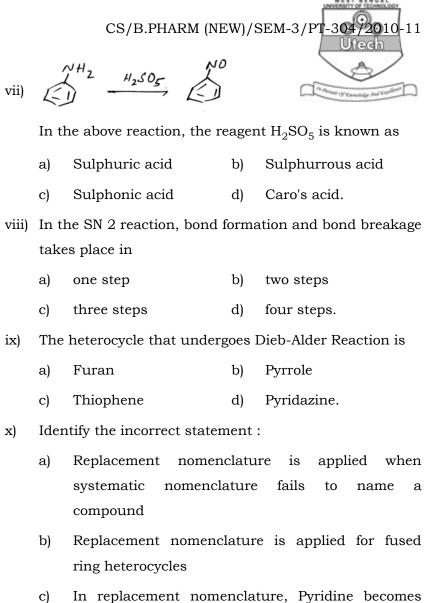
v) Haworth synthesis is employed for the preparation of

- a) Naphthalene b) Anthracene
- c) Phenanthrene d) Both (a) and (c).

vi) The following aromatic nitrocompound in an explosive is

- a) o-nitrotoluene
- b) p-nitrotoluene
- c) 2, 4 dinitrotoluene
- d) 2, 4, 6 trinitrotoluene.

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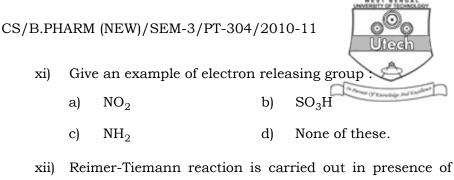


- 1-aza benzene
- Replacement nomenclature d) system solves complications arise due to Hantzsch - Widman nomenclature.

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which of the fallowing reagents ?

-)	∧ II NIINII	1_)	IION + IIOI
ai	$C_6H_5NHNH_2$	D1	HCN + HC1
a,	0615111112	~)	11011 1101

c)  $CHCl_3 + KOH$  d)  $LiAlH_4$ .

#### **GROUP – B**

## ( Short Answer Type Questions )

Answer any *three* of the following.  $3 \times 5 = 15$ 

2. What is Huckel's rule of Aromaticity ? What is the reason behind the aromaticity of pyrrole ?  $1\frac{1}{2} + 3\frac{1}{2}$ 

#### 3. a) Why is aniline less basic than aliphatic amines ?

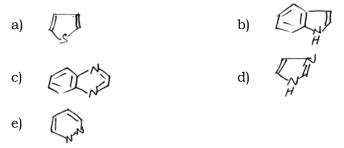
b) Write the reagents and the mechanism involved in the following reaction : 2 + (1 + 2) $\int^{N_{H_{2}}} N = N - C I$ 

 $\xrightarrow{N=N-Cl}$ 

- 4. Explain any *two* of the following :  $2 \times 2\frac{1}{2}$ 
  - a) Pyrrole is less basic than pyridine.
  - b) Epoxides are not considered as heterocyclic compound
  - c) Electrophilic substitution is preferred at -2 position in comparison to 3 position in furan.

- "Although chlorine acts as a deactivator, chlorobenzene acts as an ortho-paradirector" Explain.
- 6. Identify the structures :





#### **GROUP - C**

### (Long Answer Type Questions)

Answer any *three* of the following.  $3 \times 15 = 45$ 

7. What do you mean by heterocyclic structure ? Write down the structure of naphthalene and anthracene. Describe the aromatic character of anthracine. Write down the reduction, sulphonation and nitration reaction of anthracene. Describe the Howarth synthesis of anthoracene. Write down the name of the heterocyclic rings present in nucleic acid.

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$$1 + 1 + \frac{11}{2} + 6 + 4 + \frac{11}{2}$$

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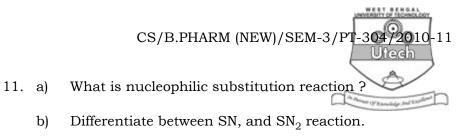
- 8. a) Explain the stepwise mechanism involved in nitration of benzene.
  - b) Briefly discuss the mechanism involved in Friedel Craft Acylation of Bengene.
  - c) Why are the functional groups WH<sub>2</sub>, OH ortho-paraorienting whereas functional groups like – NO<sub>2</sub>, – CHO are meta-orienting ?

d) Write five reactions of phenol. 
$$3 + 3 + 4 + 5$$

- Briefly describe i) Sandmeyer reaction, ii) Gatterman reaction and iii) Schiemann reaction of diazonium compounds. Write short note on hydrazines. What do you mean by leaving group ? How is Arenium ion formed during electrophilic aromatic substitution ?
- 10. a) Classify Carbohydrates with examples.
  - b) Sucrose is a non-reducing sugar. Explain with structure.
  - c) Write the chemical structure of cellulose and lactose.
  - d) Write the structures of amylose and amylopectin and point out their differences.
  - e) What do you understand by "Glycon" and "Aglycon" ?

6 + 2 + 2 + 3 + 2

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- c) Explain Mutarotation with example.
- d) Discuss about the nomenclature of fused ring
  heterocyclic systems.
  2 + 3 + 5 + 5

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