	<u>Unean</u>
Name:	(A)
Roll No.:	As Agame (V) Sample for Facilities
Invigilator's Signature :	

2012 PHARMACEUTICAL ANALYSIS

Time Allotted: 3 Hours Full Marks: 70

The figures in the margin indicate full marks.

Candidates are required to give their answers in their own words as far as practicable.

GROUP - A (Multiple Choice Type Questions)

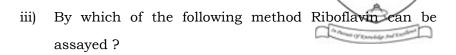
1. Choose the correct alternatives for any *ten* of the following:

 $10 \times 1 = 10$

- i) Which molecular transition require the highest energy?
 - a) $\sigma \rightarrow \sigma^*$

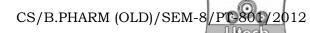
- b) $\Pi \rightarrow \Pi^*$
- c) $n. \rightarrow \Pi^*$
- d) $n \to \sigma^*$.
- ii) Why is the oxygen-hydrogen absorption of methanol such a broad band in the infrared?
 - a) Rotational energy levels broaden the absorption
 - b) Hyperconjugation resonance broadens the absorption
 - c) Resonance broadens the absorption
 - d) Hydrogen bonding broadens the absorption.

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- a) UV spectroscopy
- b) IR spectroscopy
- c) Fluorimetry
- d) Visible spectroscopy.
- iv) Glower's lamp is used as a source of radiation in
 - a) IR Spectrophotometer b) UV spectroscopy
 - c) Fluorimetry
- d) Visible spectroscopy.
- v) Stretching vibration involves
 - a) movement along the axis of the molecule
 - b) change in boad angle
 - c) movement along the axis of the bond axis
 - d) resonance.
- vi) Beer's law states that intensity of a monochromatic light decreases exponentially with the
 - a) decrease in concentration
 - b) increase in concentration
 - c) decrease in path length
 - d) increase in path length.

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- vii) Which factors influence chemical shift?
 - a) Inductive effect
- b) Anisotropic effect
- c) Hydrogen bonding
- d) All of these.
- viii) Why is the oxygen-hydrogen absorption of CH_3OH such a broad band in the infrared ?
 - a) Rotational energy levels broaden the absorption
 - b) Hyperconjugation resonance broadens the absorption
 - c) Resonance broadens the absorption
 - d) Hydrogen bonding broadens the absorption.
- ix) Which region is called the finger print region?
 - a) $4000 2500 \text{ cm}^{-1}$
- b) $2500 1400 \text{ cm}^{-1}$
- c) $1400 600 \text{ cm}^{-1}$
- d) $600 200 \text{ cm}^{-1}$.
- x) Ion separator is a component of Spectrometer?
 - a) UV-visible Spectrometer
 - b) Mass Spectrometer
 - c) IR Spectrometer
 - d) NMR Spectrometer.
- xi) Chemical shift in NMR Spectroscopy is expressed in
 - a) cm^{-1}

b) cm

c) nm

d) ppm.

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GROUP - B

(Short Answer Type Questions)

Answer any *three* of the following. $3 \times 5 = 15$

- 2. i) Give the wavelength ranges of UV, visible and infrared regions.
 - ii) What is chromophore?
 - iii) Which type of organic compounds show $\sigma \to \sigma^*$ transitions? 2+1+2
- 3. i) How will you characterize $n o \Pi^*$ transitions?
 - ii) Why is carotene a coloured compound?
 - iii) Which factors are responsible for the deviation from Beer's law ? $2+1\frac{1}{2}+1\frac{1}{2}$
- 4. i) What is molar absorption coefficient?
 - ii) What is triplet state?
 - iii) What are the differences between fluorescence and phosphorescence? $1\frac{1}{2}+1\frac{1}{2}+2$
- 5. i) What are the basic parts of a Visible Spectrophotometer?

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- ii) Why is the temperature and humidity of the room in which IR Spectrophotometer kept should be controlled?
- iii) What is stretching vibration?

$$1\frac{1}{2} + 1\frac{1}{2} + 2$$

6. Write short notes on TQM.

GROUP - C

(Long Answer Type Questions)

Answer any *three* of the following. $3 \times 15 = 45$

- 7. i) What is the advantage of gratings over prism monochromator?
 - ii) What is the principle of Fourier Transform system?
 - iii) What is Fermi resonance?
 - iv) Define the terms 'chromophore' and 'auxochrome' with examples.
 - v) What are the applications of UV Spectrophotometry?

$$2 + 3 + 3 + 4 + 3$$

- 8. i) Explain the following terms:
 - a) Spin-Spin Splitting
 - b) Chemical Shift
 - c) Sheilding mechanism.

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- ii) What are the advantages of TMS as a reference compound to be used in NMR Spectroscopy?
- iii) Name two organic compounds which can be used as solvents in NMR Spectroscopy. $(4 \times 3) + 1 + 2$
- 9. i) What are hyperchromic and hypsochromic shift?
 - ii) Define wave number.
 - iii) Which factors affect the vibrational frequency?
 - iv) Discuss the basic principle of mass spectroscopy.

$$2\frac{1}{2} + 2\frac{1}{2} + 1 + 5 + 4$$

- 10. i) What is validation?
 - ii) Discuss different types of validation.
 - iii) Give the requirements to be fulfilled for the compliance of GLP as per the stipulations provided by the regulatory authority.
 - iv) What is ISO 9000?

1 + 3 + 8 + 3

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- 11. i) Give the basic principle behind the fluorimetric assay.
 - ii) What is the significance of Fingerprint region in IR Spectroscopy?
 - iii) What are the limitations of flame photometer?
 - iv) How will you estimate riboflavin using photofluorimeter?

3 + 4 + 4 + 4

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